

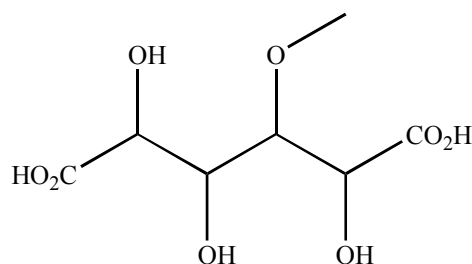
**INTEGRATING GC/MS METABOLITE PROFILING OF LATEX FROM THE
NICKEL-HYPERACCUMULATING TREE *SEBERTIA ACUMINATA* WITH LC/MS TO
TARGET THE IDENTIFICATION OF NEW Ni²⁺-COMPLEXES.**

Damien L. Callahan¹, Ute Roessner², Alan J.M. Baker³, Spas D. Kolev¹, Richard A.J. O'Hair^{1,4}

1. School of Chemistry, The University of Melbourne, Victoria, 3010, Australia. 2. Australian Centre for Plant Functional Genomics, School of Botany, The University of Melbourne, Victoria, 3010, Australia. 3. School of Botany, The University of Melbourne, Victoria, 3010, Australia 4. Bio21 Molecular Science and Biotechnology Institute, The University of Melbourne, Victoria, Australia

Sebertia acuminata Pierre ex Baill. (Sapotaceae) is a rare rainforest tree endemic to New Caledonia. It belongs to a diverse group of plants known as hyperaccumulators which store metal ions in their above-ground tissues at levels which are toxic to other species. Nickel concentrations in the latex of *S. acuminata* have been found up to 26% dry weight, the highest levels found in any organism [1]. The ligands which bind metal ions in hyperaccumulators have not been well characterised. Previous studies of the *S. acuminata* latex have reported that between 37-99.4% of the Ni is complexed by citric acid [2, 3].

The present work is a detailed phytochemical study of the latex of *S. acuminata* using LC-MS (Ion trap and FT-MS) and metabolite profiling based on GC-MS. A new Ni²⁺-complex [Ni²⁺.L₂]⁺ (L=3-methylether-2,4,5-hydroxyhexane-1,6-dioic acid, below) was identified in the latex, the first time that this organic acid has been isolated directly from biological material. More than 120 compounds were detected in the latex by GC-MS. The gas chromatography revealed that while citric acid is present in the highest concentration a number of other organic acids, subsequently found bound to Ni²⁺, were also present in high relative concentrations. A list of metabolites identified in the GC-MS of the latex was used to identify the following Ni²⁺-complexes in the LC-MS profile: Ni²⁺-citric acid [Ni(C₆H₈O₇)(C₆H₇O₇)]⁺, Ni²⁺-malic acid [Ni(C₄H₆O₅)₂(C₄H₅O₅)]⁺, Ni²⁺-itaconic acid [Ni(C₅H₅O₄)(CH₃CN)₂]⁺, Ni²⁺-erythronic acid [Ni(C₄H₈O₅)₂(C₄H₇O₅)]⁺, Ni²⁺-galacturonic acid [Ni(C₆H₁₀O₇)₂(C₆H₉O₇)]⁺, Ni²⁺-tartaric acid [Ni(C₄H₅O₆)]⁺, Ni²⁺-aconitic acid [Ni(C₆H₆O₆)₂(C₆H₅O₆)]⁺ and Ni²⁺-saccharic acid [Ni(C₆H₁₀O₈)(C₆H₉O₈)]⁺. The results are consistent with a number of organic acids having a potential role in the storage of Ni²⁺ ions within *S. acuminata*.



1. Jaffre T, Brooks RR, Lee J, Reeves RD (1976) *Science* (Washington, DC, United States) 193:579-580
2. Sagner S, Kneer R, Wanner G, Cosson JP, Deus-Neumann B, Zenk MH (1998) *Phytochemistry* 47:339-347
3. Schaumlöffel D, Ouerdane L, Bouyssiere B, Lobinski R (2003) *J Anal At Spectrom* 18:120-127